



EXAM PAPERS PRACTICE

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Detailed mark scheme

Suitable for all boards

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2002

XVIII

1583

Time allowed
70 Minutes

Score

/58

Percentage

%

CHEMISTRY

**OCR
AS & A LEVEL**

Mark Scheme

**Module 6: Organic chemistry
and analysis**

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Question			Answer	Mark	Guidance
1	(a)	(i)	 —NH ₃ ⁺ in second product ✓	3	ALLOW correct structural OR displayed OR skeletal formulae OR a combination of above as long as unambiguous ALLOW —O ⁻ Na ⁺ OR —O ⁻ (cation not required) DO NOT ALLOW —O—Na (covalent bond) DO NOT ALLOW —O (without the sodium) ALLOW delocalised carboxylate
		(ii)	perfume/fragrance/flavouring ✓	1	IGNORE solvent OR food additive
		(iii)	Reaction 3: (hot) ethanolic ammonia ✓ Reaction 4: oxidation ✓ Reaction 5: hydrolysis ✓	3	ALLOW NH ₃ (dissolved) in ethanol IGNORE other conditions ALLOW oxidisation/oxidised DO NOT ALLOW redox ALLOW nucleophilic addition-elimination DO NOT ALLOW nucleophilic substitution IGNORE acid/base



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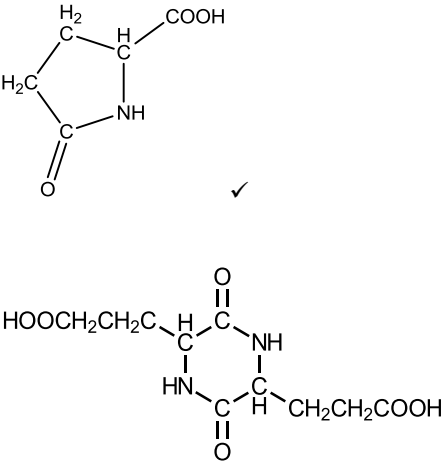
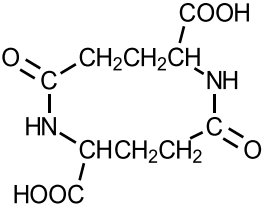
Question	Answer	Mark	Guidance
(b)	<p>M1 Compound E</p> $\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{H}_2\text{C}=\text{C}-\text{C}-\text{CHO} \\ \\ \text{NH}_2 \end{array}$ <p>✓</p> <p>M2 Compound F</p> $\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{H}_2\text{C}=\text{C}-\text{C}-\text{COOH} \\ \\ \text{NH}_2 \end{array}$ <p>✓</p> <p>M3 Compound G</p> $\left[\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ -\text{C}-\text{C}- \\ \quad \\ \text{H} \quad \text{CHNH}_2 \\ \quad \quad \\ \quad \quad \text{COOH} \end{array} \right]_n$ <p>✓</p> <p>M4 Compound H</p> $\left[\begin{array}{c} \text{H} \quad \text{O} \\ \quad \\ -\text{N}-\text{C}-\text{C}- \\ \quad \\ \text{H} \quad \text{CH}=\text{CH}_2 \end{array} \right]_n$ <p>✓</p>	6	<p>ANNOTATE ANSWER WITH TICKS AND CROSSES ETC</p> <p>ALLOW correct structural OR displayed OR skeletal formulae OR a combination of above as long as unambiguous</p> <p>Labels are not required for compound E, F, G or H</p> <p>IGNORE labels for M1, M2, M3 and M4</p> <p>CH₂=CH must be shown in E</p> <p>ALLOW C₂H₃ OR CHCH₂ for CH=CH₂ in F</p> <p>ALLOW ECF from error in structure of <u>aldehyde E</u></p> <p>ALLOW multiple repeat units but must be full repeat units</p> <p>ALLOW end bonds shown as</p> <p>DO NOT ALLOW if structures have no end bonds</p> <p>IGNORE brackets unless they are used to pick out the repeat unit from a polymer chain</p> <p>IGNORE n</p> <p>ALLOW C₂H₄NO₂ for CH(NH₂)COOH in polymer G</p> <p>ALLOW C₂H₃ OR CHCH₂ for CH=CH₂ in polymer H</p> <p>ALLOW ECF from NH₂CH₂CH=CHCOOH for the formation of compound G or compound H</p>



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Question	Answer	Mark	Guidance
	<p>M5 Compound G OR</p> $\left[\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{---C---C---} \\ \quad \\ \text{H} \quad \text{CHNH}_2 \\ \quad \quad \\ \quad \quad \text{COOH} \end{array} \right]_n$ <p>Is an addition polymer ✓</p> <p>M6 Compound H OR</p> $\left[\begin{array}{c} \text{H} \quad \text{O} \\ \quad \\ \text{---N---C---C---} \\ \quad \\ \text{H} \quad \text{CH=CH}_2 \end{array} \right]_n$ <p>is a condensation polymer ✓</p>		<p>ALLOW alkene forms addition polymer/polymer with same empirical formula as monomer</p> <p>ALLOW equation for reaction</p> $n \text{ H}_2\text{C}=\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{---C---C---} \\ \quad \\ \text{H} \quad \text{NH}_2 \end{array} \text{COOH} \rightarrow \left[\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{---C---C---} \\ \quad \\ \text{H} \quad \text{CHNH}_2 \\ \quad \quad \\ \quad \quad \text{COOH} \end{array} \right]_n$ <p>ALLOW amino acid forms condensation polymer</p> <p>OR (molecules of) compound F join/bond/add/react/form polymer and water/small molecule</p> <p>ALLOW equation for reaction</p> $n \text{ H}_2\text{C}=\begin{array}{c} \text{H} \quad \text{H} \\ \quad \\ \text{---C---C---} \\ \quad \\ \text{H} \quad \text{NH}_2 \end{array} \text{COOH} \rightarrow \left[\begin{array}{c} \text{H} \quad \text{O} \\ \quad \\ \text{---N---C---C---} \\ \quad \\ \text{H} \quad \text{CH=CH}_2 \end{array} \right]_n + \text{H}_2\text{O}$
(c) (i)	$\begin{array}{ccccccc} & \text{H} & \text{H} & \text{H} & \text{H} & \text{O} & \\ & & & & & & \\ \text{H} & \text{---N---} & \text{C---} & \text{C---} & \text{C---} & \text{C---} & \text{OH} \\ & & & & & & \\ & & \text{HOOC} & \text{H} & \text{H} & & \end{array}$ <p style="text-align: right;">✓</p>	1	<p>ALLOW correct structural OR displayed OR skeletal formulae</p> <p>OR a combination of above as long as unambiguous</p>



Question	Answer	Mark	Guidance
(ii)	 <p>✓</p> <p>✓</p>	2	<p>ALLOW correct structural OR displayed OR skeletal formulae OR a combination of above as long as unambiguous</p> <p>ALLOW a cyclic amide with a 3 membered ring</p> <p>ALLOW</p>  <p>OR a structure obtained by condensation of a glutamic acid molecule with the first cyclic amide</p>

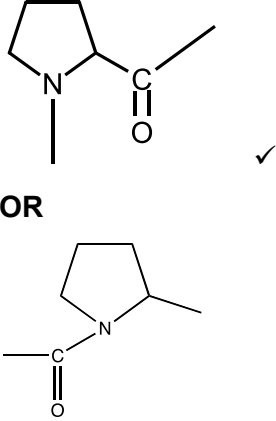
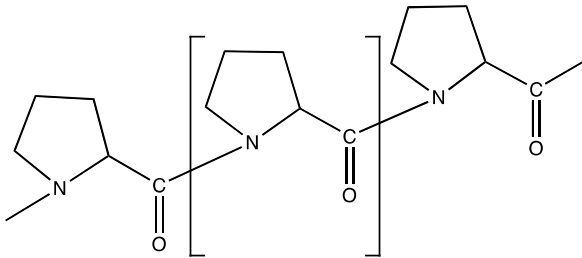


Question		Answer	Mark	Guidance
(d)	(i)	Ester AND amide ✓	1	ALLOW peptide for amide
	(ii)	 	2	ALLOW correct structural OR displayed OR skeletal formulae OR a combination of above as long as unambiguous Functional groups do not need to be fully displayed ALLOW structures as shown; the O-H bond and the N-H bonds in the functional groups do not need to be displayed DO NOT ALLOW -COOH ALLOW
	(iii)	(The molecule/amide/ester) can be <u>hydrolysed</u> ✓	1	ALLOW (the molecule/amide/ester) can form hydrogen/H-bonds <u>with water</u> IGNORE acid/base
Total			20	

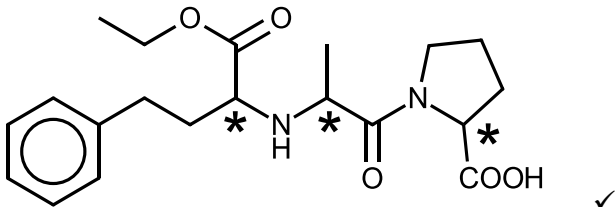


Question			Answer	Mark	Guidance
2	(a)	(i)	$\begin{array}{ccccccc} & \text{H} & \text{O} & & \text{CH}_2\text{OH} & & \\ & & & & & & \\ \text{H}_2\text{N} & -\text{C} & -\text{C} & -\text{N} & -\text{C} & -\text{COOH} \\ & & & & & & \\ & \text{CH}_3 & & \text{H} & \text{H} & & \end{array}$ <p style="text-align: right;">✓</p> $\begin{array}{ccccccc} & \text{H} & \text{O} & & \text{CH}_3 & & \\ & & & & & & \\ \text{H}_2\text{N} & -\text{C} & -\text{C} & -\text{N} & -\text{C} & -\text{COOH} \\ & & & & & & \\ & \text{HOH}_2\text{C} & & \text{H} & \text{H} & & \end{array}$ <p style="text-align: right;">✓</p>	2	ALLOW correct structural OR displayed OR skeletal formulae OR combination of above as long as unambiguous DO NOT ALLOW peptide chains
	(a)	(ii)	alanine at pH 6.0 $\begin{array}{ccc} & \text{H} & \text{O} \\ & & \\ \text{H}_3\text{N}^{\oplus} & -\text{C} & -\text{C} & -\text{O}^{\ominus} \\ & & & \\ & \text{CH}_3 & & \end{array}$ <p style="text-align: right;">✓</p> serine at pH 10.0 $\begin{array}{ccc} & \text{H} & \text{O} \\ & & \\ \text{H}_2\text{N} & -\text{C} & -\text{C} & -\text{O}^{\ominus} \\ & & & \\ & \text{CH}_2\text{OH} & & \end{array}$ <p style="text-align: right;">✓</p>	2	ALLOW correct structural OR displayed OR skeletal formulae OR combination of above as long as unambiguous ALLOW + charge on N or H: <i>i.e.</i> $^+\text{NH}_3$ or NH_3^+ DO NOT ALLOW '- ' charge on C <i>i.e.</i> ^-COO DO NOT ALLOW if structure is incomplete

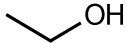
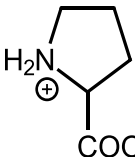
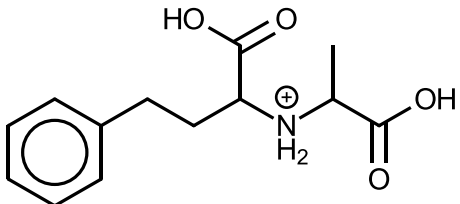




Question	Answer	Mark	Guidance
(a) (iii)	 <p>OR</p>	1	<p>ALLOW correct structural OR displayed OR skeletal formulae OR combination of above as long as unambiguous</p> <p>IGNORE bond angles</p> <p>DO NOT ALLOW more than one repeat unit</p> <p>ALLOW end bonds shown as - - - -</p> <p>DO NOT ALLOW if structure has no end bonds</p> <p>IGNORE brackets unless they are used to pick out the repeat unit from a polymer chain</p> <p>IGNORE n</p> 

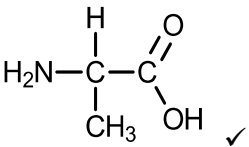
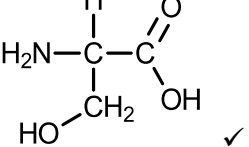
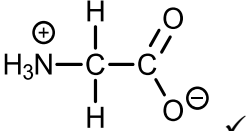
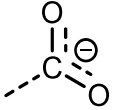


Question		Answer	Mark	Guidance												
(b)		<table border="1"><thead><tr><th colspan="3">^1H NMR spectrum for serine</th></tr><tr><th>chemical shift, δ /ppm</th><th>relative peak area</th><th>splitting pattern</th></tr></thead><tbody><tr><td>2.0 to 3.0</td><td>1</td><td>triplet</td></tr><tr><td>3.3 to 4.2</td><td>2</td><td>doublet</td></tr></tbody></table> <p>One mark for each correct row ✓✓</p>	^1H NMR spectrum for serine			chemical shift, δ /ppm	relative peak area	splitting pattern	2.0 to 3.0	1	triplet	3.3 to 4.2	2	doublet	2	<p>ALLOW δ values ± 0.2 ppm, as a range or a value within the range</p> <p>ALLOW a response that implies a splitting into three for a triplet/into two for a doublet</p>
^1H NMR spectrum for serine																
chemical shift, δ /ppm	relative peak area	splitting pattern														
2.0 to 3.0	1	triplet														
3.3 to 4.2	2	doublet														
(c)	(i)		1	ALL correct for one mark												
(c)	(ii)	<p>any two from:</p> <p>no/fewer side effects</p> <p>increases the (pharmacological) activity/effectiveness</p> <p>Reduces/stops the need for/cost/difficulty in separating stereoisomers/optical isomers</p> <p>✓✓</p>	2	<p>IGNORE toxic/harmful</p> <p>IGNORE a response that implies a reduced dose</p> <p>IGNORE "it takes (less) time to separate"</p>												



Question		Answer	Mark	Guidance
(c)	(iii)	<p> ✓ one mark for ethanol</p> <p> ✓ one mark for proline with NH OR NH₂⁺</p> <p></p> <p>✓ one mark for remaining fragment</p> <p>with  or </p> <p>✓ Fourth mark for structure of both ions shown correctly with NH₂⁺</p>	4	<p>ALLOW correct structural OR displayed OR skeletal formulae</p> <p>OR combination of above as long as unambiguous</p> <p>ALLOW + charge on H of NH₂ groups, <i>i.e.</i> NH₂⁺</p> <p>IGNORE negative (counter) ions</p>
(c)	(iv)	<p>idea of separating (the components/compounds)</p> <p>AND idea of (identifying compounds by) comparison with a (spectral) database ✓</p>	1	<p>ALLOW (identifies compounds) using fragmentation (patterns)/fragment ions (but IGNORE molecular ions)</p> <p>IGNORE retention times</p>
Total			15	



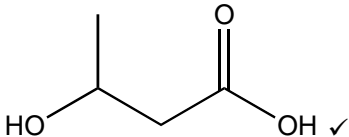
Question			Answer	Marks	Guidance
3	(a)	(i)	monomers join/bond/add/react/form polymer/form chain AND another product/small molecule e.g. H ₂ O/HCl ✓	1	IGNORE 'two' when referring to monomers, <i>i.e.</i> (two) monomers...
		(ii)	 ✓  ✓	2	ALLOW correct structural OR displayed OR skeletal formulae OR combination of above as long as unambiguous ALLOW zwitterions
		(iii)	The pH at which the zwitterion exists ✓  ✓	2	ALLOW correct structural OR displayed OR skeletal formulae OR combination of above as long as unambiguous ALLOW pH at which there is no overall/net charge IGNORE pH at which there is no charge/ neutral charge <i>ie overall/net is required</i> ALLOW pH at which contains COO ⁻ AND NH ₃ ⁺ 
	(b)	(i)	Adsorption ✓	1	DO NOT ALLOW absorption ALLOW partition ALLOW adsorbtion
		(ii)	R _f = 0.53 to 0.62 ✓ Amino acid is <u>methionine</u> ✓	2	Values vary if distance measured to middle or top of spot Independent marks. No need to show working as question asks for estimate of R _f



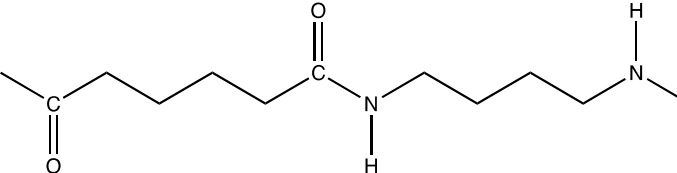
Question		Answer	Marks	Guidance
(c)		<p>amide link ✓ correct structure ✓</p>	2	ALLOW correct structural OR displayed OR skeletal formulae OR combination of above as long as unambiguous ALLOW 'terminal' —NH— at other end 'End bonds' MUST be shown (solid or dotted) IGNORE brackets and/or <i>n</i> DO NOT ALLOW aromatic rings in amine residue ALLOW CONH for amide link
(d)	(i)	<p>Penalise connectivity once (i.e. not -HO)</p>	2	ALLOW correct structural OR displayed OR skeletal formulae OR combination of above as long as unambiguous For dicarboxylic acid: ALLOW diethyl chloride DO NOT ALLOW the CIS monomer
	(ii)		1	ALLOW correct structural OR displayed OR skeletal formulae OR combination of above as long as unambiguous
Total			13	



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Question			er	Marks	Guidance
4	(a)	(photodegradable OR light/sunlight/UV ✓	1	IGNORE IR/heat IGNORE bacteria DO NOT ALLOW burn/combustion
		(ii)	 HO-CH(CH ₃)-CH ₂ -C(=O)OH ✓	1	DO NOT ALLOW structure with any C shown (especially as part of C=O) DO NOT ALLOW OH—
	(b)	(ammonia/NH ₃ AND ethanol OR ethanolic ammonia ✓	1	ALLOW ammonia in a sealed tube IGNORE heat ALLOW dilute ethanolic ammonia /NH ₃ DO NOT ALLOW any reference to water or hydroxide ions, e.g. DO NOT ALLOW dilute ethanolic NH ₃ (aq) e.g. DO NOT ALLOW ethanolic NH ₃ + NaOH
		(ii)	Nitrogen electron pair/lone pair accepts a proton/H ⁺ ✓ <i>Requires position of electron pair on N</i> Cl ⁻ H ₃ N ⁺ (CH ₂) ₄ N ⁺ H ₃ Cl ⁻ OR ClH ₃ N(CH ₂) ₄ NH ₃ Cl ✓	2	DO NOT ALLOW Nitrogen/N lone pair accepts hydrogen <i>proton/H⁺ required</i> ALLOW nitrogen donates an electron pair IGNORE NH ₂ group donates electron pair ALLOW + charge (if shown) on N or H of NH ₃ e.g. Cl ⁻ H ₃ N ⁺ (CH ₂) ₄ NH ₃ ⁺ Cl ⁻ DO NOT ALLOW just H ₃ N ⁺ (CH ₂) ₄ NH ₃ ⁺ i.e. 2 x Cl ⁻ MUST be included



Question	er	Marks	Guidance
	<p>(iii) 1 mark for amide/peptide link correctly displayed within an attempted repeat unit ✓</p> <p>1 mark for rest of structure correct including side links ✓</p> $\begin{array}{ccccccc} & \text{O} & & \text{O} & & & \\ & & & & & & \\ \text{---} & \text{C} & \text{---} & (\text{CH}_2)_4 & \text{---} & \text{C} & \text{---} \\ & & & & & & \\ & & & & & \text{H} & \\ & & & & & & \\ & & & & & & \\ & & & & & \text{H} & \\ & & & & & & \\ & & & & & \text{H} & \end{array}$	2	<p>Minimum requirement is each end of a displayed amide group attached to a carbon atom (could be skeletal)</p> <p>Brackets not required</p> <p>IF more than one repeat unit has been drawn a single repeat unit MUST be identified by brackets or clear label</p> <p>DO NOT ALLOW 2nd mark if amide/peptide link wrong <i>1st mark requires amide group fully displayed</i> <i>For 2nd mark, ALLOW –CONH– in correct structure</i></p> <p>ALLOW correct structural OR displayed OR skeletal formula</p> <p>ALLOW combination of formulae as long as unambiguous e.</p> 



Question		er	Marks	Guidance
(c)	(i)	<p>One mark for each correct structure</p> $\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_3\text{N}^+ - \text{CH} - \text{C} - \text{O}^- \\ \\ \text{CHOH} \\ \\ \text{CH}_3 \end{array} \quad \checkmark$ $\begin{array}{c} \text{O} \\ \parallel \\ \text{H}_3\text{N}^+ - \text{CH} - \text{C} - \text{O}^- \\ \\ (\text{CH}_2)_4 \\ \\ \text{NH}_2 \end{array} \quad \text{OR} \quad \begin{array}{c} \text{O} \\ \parallel \\ \text{H}_2\text{N} - \text{CH} - \text{C} - \text{O}^- \\ \\ (\text{CH}_2)_4 \\ \\ \text{NH}_3^+ \end{array} \quad \checkmark$	2	<p>ALLOW correct structural OR displayed OR skeletal formula ALLOW combination of formulae as long as unambiguous</p> <p>ALLOW COO⁻ '-' charge must be on O of COO⁻ but</p> <p>ALLOW + sign shown as ⁺NH₃ OR NH₃⁺ BUT only one NH₂ can be protonated in zwitterion</p>
	(ii)	<p>Zwitterion at pH 9.60/higher pH has one NH₂ group OR Zwitterion OR amino acid at pH 9.60/higher pH has a side chain with an NH₂ group ✓</p> <p>Note: ASSUME that 'it' refers to zwitterion</p>	1	<p>ALLOW amino acid at 9.60/higher pH has two NH₂ groups ALLOW amino acid at 9.60/higher pH has more NH₂ groups ALLOW amine OR amino for NH₂ IGNORE CHOH slightly acidic</p>
Total			10	