

Proteins and enzymes 3

Level: Edexcel A Level 9BN0

Subject: Biology

Exam Board: Suitable for all boards

Topic: Proteins and enzymes 3

Type: Questionnaire

To be used by all students preparing for Edexcel Biology A Level 9BN0 foundation or higher tier but also suitable for students of other boards.



1	Uric acid is produced in the body. One of the reactions involved in the production of uric acid is
١	catalysed by xanthine oxidase.

	xanthine oxidase	
	xanthine ———► uric acid	
(a)	A sample of xanthine oxidase was tested by mixing with biuret reagent. Describe and explain the result of this test.	
		(2)
(b)	Explain why xanthine oxidase is able to catalyse this reaction but it is not able to catalyse other reactions.	. ,
		(2)
(c)	Gout is a painful condition caused by uric acid crystals in the joints. It is often treated wirding that inhibits xanthine oxidase. The diagram shows a molecule of xanthine and a molecule of this drug.	th a

Xanthine Drug used to treat gout



Use the diagram to explain why this drug is effective in the treatment of gout.	
	_
	_
	_
	_
	_
	_
	(3)
	(Total 7 marks)

For more help visit our website https://www.exampaperspractice.co.uk/



2 Lactose is a disaccharide found in milk. In the human small intestine, the enzyme lactase catalyses the hydrolysis of lactose to the monosaccharides, galactose and glucose. These monosaccharides are then absorbed into the blood.

SComplete the diagram to show the hydrolysis of lactose to galactose and glucose.

(Total 2 marks)

- The box jellyfish produces a poison (venom) which enters the blood when a person is stung. A person who has been stung can be treated with an injection of antivenom. This antivenom is produced by injecting small amounts of venom from box jellyfish into sheep, then extracting antibodies from the sheeps' blood. These antibodies are then injected into the person who has been stung.
 - (a) If a sheep is injected with the box jellyfish venom on more than one occasion a higher yield of antivenom is obtained. Explain why.



(b)	Injecting antivenom does not give a person lasting protection against the venom of jellyfish. Explain why.	box
		_ _ _
(c)	Suggest one possible problem in injecting people with antivenom made in this way	(2)
		— (1) (Total 5 marks)



_	
1	

(a) The table contains statements about three stages of respiration.

Complete the table with a tick if the statement in the first column is true for each stage of respiration in an animal.

	Glycolysis	Link reaction	Krebs cycle
Occurs in mitochondria			
Carbon dioxide produced			
NAD is reduced			

(b)	The following reaction occurs in the Krebs cycle.	
	Enzyme	
	Succinate —	Fumarate

A scientist investigated the effect of the enzyme inhibitor malonate on this reaction. The structure of malonate is very similar to the structure of succinate. The scientist added malonate and the respiratory substrate, pyruvate, to a suspension of isolated mitochondria. She also bubbled oxygen through the suspension.

				
				 -
explain how malo	nate inhibits the form	mation of fumara	te from succina	te.

(3)



		(iii)			by the mitochondria during ed when malonate was add	
						(2) (Total 9 marks)
5	(a)	Nam	e the monosacchar	ides of which the following	disaccharides are compose	d.
J		(i)	Sucrose			
			monosaccharides	and	d	
						(1)
		(ii)	Lactose			
			monosaccharides	and	<u> </u>	(1)
	(b)	Amy	/lase and maltase a	re involved in the digestion	of starch in the small intest	
	()	Con		dentifying where these enzy	mes are produced and the	
		N	Name of enzyme	Where the enzyme is produced	Product of the reaction catalysed by the enzyme	
			Amylase			
			Maltase			

(2) (Total 4 marks)

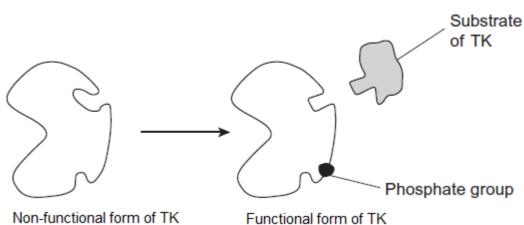


6

The enzyme tyrosine kinase (TK) is found in human cells. TK can exist in a non-functional and a functional form. The functional form of TK is only produced when a phosphate group is added to TK.

This is shown in Figure 1.

Figure 1



(a) Addition of a phosphate group to the non-functional form of TK leads to production of the functional form of TK.

Explain how.			



(b) The binding of the functional form of TK to its substrate leads to cell division. Chronic myeloid leukaemia is a cancer caused by a faulty form of TK. Cancer involves uncontrolled cell division.

Figure 2 shows the faulty form of TK.

Figure 2



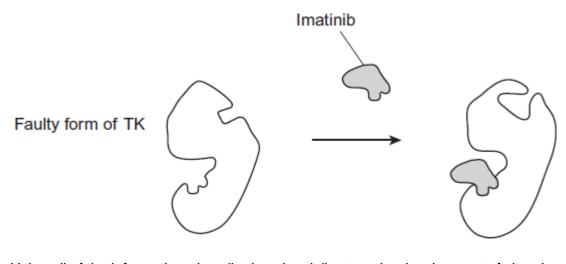
Faulty form of TK

suggest now faulty 1 K leads to chronic myeloid leukaemia.					



(c) Imatinib is a drug used to treat chronic myeloid leukaemia. **Figure 3** shows how imatinib inhibits faulty TK.

Figure 3



Using all of the inform myeloid leukaemia.	nation, describe how	imatinib stops the o	development of chron	IC

(2)

(Total 6 marks)



Mycolic acids are substances that form part of the cell wall of the bacterium that causes 7 tuberculosis. Mycolic acids are made from fatty acids. Isoniazid is an antibioticthat is used to treat tuberculosis. The diagram shows how this antibiotic inhibits the production of mycolic acids in this bacterium. Isoniazid (inactive) enzyme B Isoniazid (active) inhibits substrates fatty acids mycolic acids (a) Treatment with isoniazid leads to the osmotic lysis of this bacterium. Use information in the diagram to suggest how. (2) (b) Human cells also produce fatty acids. Isoniazid does not affect the production of these fatty acids. Use information in the diagram to suggest one reason why isoniazid does not affect the production of fatty acids in human cells.

(1)

(c)		utation in the gene coding for enzyme B could lead to the production of a -functional enzyme. Explain how.	
			_
			_
	(Ext	tra space)	_
The	diagra	am shows the structure of a bacterium and the sites of action of two antibiotics	i .
		comycin acts he cell wall on ribosomes	
(a)	(i)	Use information in the diagram to explain why vancomycin does not affect holes.	uman
			_
	(ii)	Use information in the diagram to explain how tetracycline prevents bacteria	growth.
			_
			 (1)



(b)	Fred	quent treatment with vancomycin can result in resistant strains of bacteria. Explair	how.
	(Ext	tra space)	
			(2)
		(To	otal 4 marks)
The	equat	tion shows the breakdown of lactose by the enzyme lactase.	
		la ctase	
Lact	ose +	· water galactose + monosaccharide X	
(a)	(i)	Name the type of reaction catalysed by the enzyme lactase.	
			(1)
	(ii)	Name monosaccharide X.	
			(1)
(b)	(i)	Describe how you would use a biochemical test to show that a reducing sugar is present.	3
			(2)

For more help visit our website https://www.exampaperspractice.co.uk/

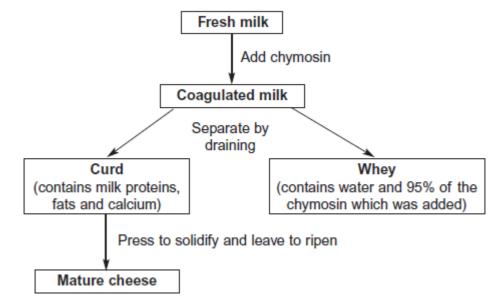


	(ii)	Lactose, galactose and monosaccharide X are all reducing sugars. After the lactose has been broken down there is a higher concentration of reduci sugar. Explain why.	ing
(c)		A high concentration of galactose slows down the breakdown of lactose by lactase. Use your knowledge of competitive inhibition to suggest why.	
			(1

(Total 7 marks)



The figure below summarises the way in which cheese is made.



	······································
	······································
-	rcentage of protein in mature cheese is greater than the percentage of protein in atted milk. Use the flow chart to explain why.

(1)

(Total 3 marks)

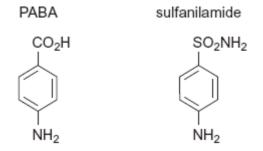
For more help visit our website https://www.exampaperspractice.co.uk/



11

Folic acid is a substance required by bacteria for cell growth. Bacteria produce folic acid by the following reaction.

The diagram shows the structure of a molecule of PABA. It also shows the structure of a molecule of a drug called sulfanilamide, which can be used to treat bacterial infections. Sulfanilamide prevents bacteria producing folic acid.



Use the diagram and your knowledge of enzymes to explain how sulphanilamide prevent bacteria producing folic acid.	S
	_
	_
	_
	_
	_
	_
	(Total 3 marks)



12

(a) The table shows some substances found in cells. Complete the table to show the properties of these substances. Put a tick in the box if the statement is correct.

	Substance					
Statement	Starch	Glycogen	Deoxyribose	DNA helicase		
Substance contains only the elements carbon, hydrogen and oxygen						
Substance is made from amino acid monomers						
Substance is found in both animal cells and plant cells						

(4)

(b) The diagram shows two molecules of β -glucose.

(ii)

On the diagram, draw a box around the atoms that are removed when the two β -glucose molecules are joined by condensation.

(2)

(c) (i) Hydrogen bonds are important in cellulose molecules. Explain why.

(2)

function in cells.			

A starch molecule has a spiral shape. Explain why this shape is important to its

(1)

(Total 9 marks)



Hexosaminidase only breaks down gangliosides. It does not break down other lipids.

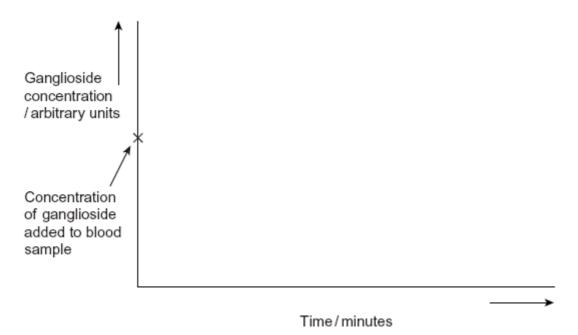
13	Gangliosides are lipids found in the cell surface membranes of nerve cells. Hexosaminidase is an
	enzyme present in blood that breaks down gangliosides. If gangliosides are not broken down,
	they damage nerve cells

(a)

(b) Hexosaminidase is found in the blood of healthy people. People with Tay Sachs disease do not have this enzyme in their blood.

Doctors confirm Tay Sachs disease by using a blood test. The technician carrying out the test adds a solution containing a high concentration of gangliosides to a sample of blood from the person being tested. The technician then measures the concentration of gangliosides in the person's blood at regular intervals.

(i) Complete the graph below by sketching a curve to show the results you would expect for a person with Tay Sachs disease. Label this curve **T**.



(1)

(3)



Sketch a curve on the same graph to show the results you would expect for a healthy

(ii)

(c)	Scientists are trying to find a way to give the missing enzyme to people with Tay Sac disease. Suggest why they cannot give the enzyme as a tablet that is swallowed.	chs
		_
		_
		(Total 7 r
(a)	Nitrogen gas is the usual substrate for this enzyme. Name the product.	
		_
(b)	Nitrogenase also catalyses reactions involving other substances. Explain what this suggests about the shapes of the molecules of these other substances.	_
(b)	Nitrogenase also catalyses reactions involving other substances. Explain what this	_
(b)	Nitrogenase also catalyses reactions involving other substances. Explain what this	- - -



c)	(i)	Azotobacter is a nitrogen-fixing bacterium. It produces the enzyme nitrogenase. The enzyme only works in the absence of oxygen.
		Azotobacter has a very high rate of aerobic respiration compared with bacteria that do not fix nitrogen. Suggest two advantages of the very high rate of aerobic respiration.
	(ii)	If scientists could transfer the gene that codes for nitrogenase to cereal plants, these cereal plants would be able to fix nitrogen. However, the scientists would expect these genetically engineered cereal plants to grow more slowly than cereal plants that get their nitrogen from fertiliser. Explain why they would grow more slowly.

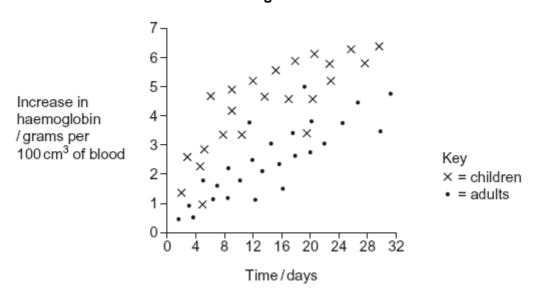
(Total 7 marks)

For more help visit our website https://www.exampaperspractice.co.uk/



(a) Haemoglobin contains iron. One type of anaemia is caused by a lack of iron. This type of anaemia can be treated by taking tablets containing iron. A number of patients were given a daily dose of 120 mg of iron. **Figure 1** shows the effect of this treatment on the increase in the concentration of haemoglobin in their red blood cells.

Figure 1



Give one difference in the response of adults and children to this treatment.
Vou could use the graph to predict the effect of this treatment on the increase in
You could use the graph to predict the effect of this treatment on the increase in haemoglobin content of an adult after 40 days. Explain how.

(iii) Haemoglobin has a quaternary structure. Explain what is meant by a quaternary structure.

(1)

(i)	Pernicious anaemia is another type of anaemia. One method of identifying pernicionanaemia is to measure the diameter of the red blood cells in a sample of blood that has been diluted with an isotonic salt solution. Explain why an isotonic salt solution used to dilute the blood sample.
(ii)	A technician compared the red blood cells in two blood samples of equal volume. One sample was from a patient with pernicious anaemia, the other was from a pat who did not have pernicious anaemia. Figure 2 shows some of the results she obtained.
	Figure 2
	Patient without pernicious anaemia
	Number of ed blood cells 60-ber unit volume 40-
	20- 0 * * * * * * * 9 10 11 12
	Diameter/μm
	Describe two differences between the blood samples.

(2)

(3)

(Total 9 marks)



Lactose + → Glucose + (ii) Name the type of chemical reaction shown in this equation. (b) Lactase is an enzyme. Lactose is a reducing sugar. (i) Describe how you could use the biuret test to distinguish a solution of the enzyme, lactase from a solution of lactose. (ii) Explain the result you would expect with the enzyme. (iii) Explain the result you would expect with the enzyme. (Total 5 A glucose biosensor is an instrument used to measure glucose concentration. It contains an enzyme called glucose oxidase. (a) A glucose biosensor detects only glucose. Use your knowledge of the way in which enzymes work to explain why.	(a)	(i)	The equation shows the reaction catalysed by the enzyme lactase. Complete this equation.
(b) Lactase is an enzyme. Lactose is a reducing sugar. (i) Describe how you could use the biuret test to distinguish a solution of the enzyme, lactase from a solution of lactose. (ii) Explain the result you would expect with the enzyme. (Total 5 A glucose biosensor is an instrument used to measure glucose concentration. It contains an enzyme called glucose oxidase. (a) A glucose biosensor detects only glucose. Use your knowledge of the way in which			Lactose +
(i) Describe how you could use the biuret test to distinguish a solution of the enzyme, lactase from a solution of lactose. (ii) Explain the result you would expect with the enzyme. (Total 5 A glucose biosensor is an instrument used to measure glucose concentration. It contains an enzyme called glucose oxidase. (a) A glucose biosensor detects only glucose. Use your knowledge of the way in which		(ii)	Name the type of chemical reaction shown in this equation.
(i) Describe how you could use the biuret test to distinguish a solution of the enzyme, lactase from a solution of lactose. (ii) Explain the result you would expect with the enzyme. (Total 5 A glucose biosensor is an instrument used to measure glucose concentration. It contains an enzyme called glucose oxidase. (a) A glucose biosensor detects only glucose. Use your knowledge of the way in which			
lactase from a solution of lactose. (ii) Explain the result you would expect with the enzyme. (Total 5 A glucose biosensor is an instrument used to measure glucose concentration. It contains an enzyme called glucose oxidase. (a) A glucose biosensor detects only glucose. Use your knowledge of the way in which	(b)	Lact	ase is an enzyme. Lactose is a reducing sugar.
(ii) Explain the result you would expect with the enzyme. (Total 5) A glucose biosensor is an instrument used to measure glucose concentration. It contains an enzyme called glucose oxidase. (a) A glucose biosensor detects only glucose. Use your knowledge of the way in which		(i)	
(Total 5) A glucose biosensor is an instrument used to measure glucose concentration. It contains an enzyme called glucose oxidase. (a) A glucose biosensor detects only glucose. Use your knowledge of the way in which			
(Total 5) A glucose biosensor is an instrument used to measure glucose concentration. It contains an enzyme called glucose oxidase. (a) A glucose biosensor detects only glucose. Use your knowledge of the way in which		(ii)	
A glucose biosensor is an instrument used to measure glucose concentration. It contains an enzyme called glucose oxidase. (a) A glucose biosensor detects only glucose. Use your knowledge of the way in which			
enzyme called glucose oxidase. (a) A glucose biosensor detects only glucose. Use your knowledge of the way in which			(Total
	(a)	_	



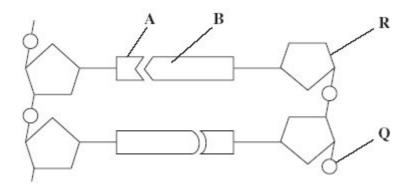
1	
2	
(i)	Diabetes mellitus is a disease that can lead to an increase in blood glucose concentration. Some diabetics need insulin injections. Insulin is a protein so it cannot be taken orally. Suggest why insulin cannot be taken orally.
(ii)	A drug company produced a new type of insulin. Scientists from the company carried out a trial in which they gave this new type of insulin to rats. They reported that the results of this trial on rats were positive. A newspaper stated that diabetics would benefit from this new drug. Suggest two reasons why this statement should be viewed with caution.
	1. 2.

(Total 8 marks)



18 Figure 1 shows a short section of a DNA molecule.

Figure 1



- (a) Name parts R and Q.
 - (i) R _____
 - (ii) **Q** _____
- (b) Name the bonds that join **A** and **B**.

(c) Ribonuclease is an enzyme. It is 127 amino acids long.

What is the minimum number of DNA bases needed to code for ribonuclease?

(1)

(2)

(1)



(d) **Figure 2** shows the sequence of DNA bases coding for seven amino acids in the enzyme ribonuclease.

Figure 2

GTTTACTACTCTTCTTTA

The number of each type of amino acid coded for by this sequence of DNA bases is shown in the table.

Amino acid	Number present
Arg	3
Met	2
Gln	1
Asn	1

Use the table and **Figure 2** to work out the sequence of amino acids in this part of the enzyme. Write your answer in the boxes below.

Gln							
Explain how enzyme.	a change in a	a sequence c	of DNA bases	could result	in a non-fund	ctional	(1
							

(3) (Total 8 marks)

(a) Sucrose, maltose and lactose are disaccharides.

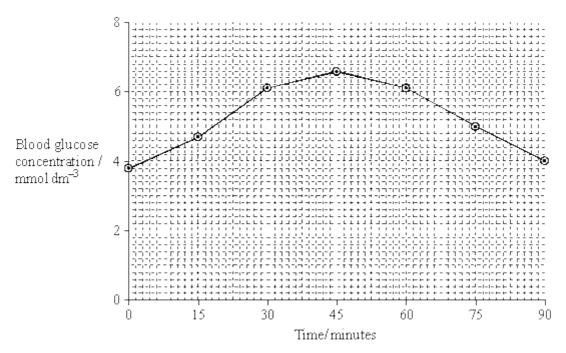
19

(i)	Sucrase is an enzyme. It hydrolyses sucrose during digestion. Name the products of this reaction.
	and



(ii)	Sucrase does not hydrolyse lactose. Use your knowledge of the way in which enzymes work to explain why.

(b) A woman was given a solution of sucrose to drink. Her blood glucose concentration was measured over the next 90 minutes. The results are shown on the graph.



(i) Describe how the woman's blood glucose concentration changed in the period shown in the graph.

(2)



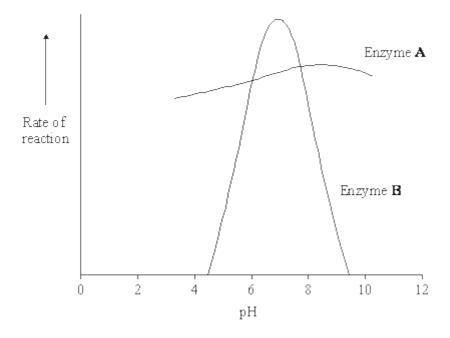
20

Some enymes digest protein. They hydrolyse the peptide bonds between amino acids. The extent to which a protein is digested is called the degree of hydrolysis (DH). The DH value may be calculated from the equation: $DH = \frac{100 \times \text{Number of peptide bonds hydrolysed}}{\text{Total number of peptide bonds present}}$ (a) (i) A protein molecule contains 151 amino acids. What is the total number of peptide bonds in this molecule? (ii) A molecule of this protein is digested. The DH value of the digested protein is 18. Calculate the number of peptide bonds that have been hydrolysed. Answer		(ii)	Explain the results shown on the graph.	
Some enymes digest protein. They hydrolyse the peptide bonds between amino acids. The extent to which a protein is digested is called the degree of hydrolysis (DH). The DH value may be calculated from the equation: $DH = \frac{100 \times \text{Number of peptide bonds hydrolysed}}{\text{Total number of peptide bonds present}}$ (a) (i) A protein molecule contains 151 amino acids. What is the total number of peptide bonds in this molecule? (ii) A molecule of this protein is digested. The DH value of the digested protein is 18. Calculate the number of peptide bonds that have been hydrolysed. Answer				
Some enymes digest protein. They hydrolyse the peptide bonds between amino acids. The extent to which a protein is digested is called the degree of hydrolysis (DH). The DH value may be calculated from the equation: $DH = \frac{100 \times \text{Number of peptide bonds hydrolysed}}{\text{Total number of peptide bonds present}}$ (a) (i) A protein molecule contains 151 amino acids. What is the total number of peptide bonds in this molecule? (ii) A molecule of this protein is digested. The DH value of the digested protein is 18. Calculate the number of peptide bonds that have been hydrolysed. Answer				
Some enymes digest protein. They hydrolyse the peptide bonds between amino acids. The extent to which a protein is digested is called the degree of hydrolysis (DH). The DH value may be calculated from the equation: $DH = \frac{100 \times \text{Number of peptide bonds hydrolysed}}{\text{Total number of peptide bonds present}}$ (a) (i) A protein molecule contains 151 amino acids. What is the total number of peptide bonds in this molecule? (ii) A molecule of this protein is digested. The DH value of the digested protein is 18. Calculate the number of peptide bonds that have been hydrolysed. Answer				(2)
extent to which a protein is digested is called the degree of hydrolysis (DH). The DH value may be calculated from the equation: $DH = \frac{100 \times \text{Number of peptide bonds hydrolysed}}{\text{Total number of peptide bonds present}}$ (a) (i) A protein molecule contains 151 amino acids. What is the total number of peptide bonds in this molecule? (ii) A molecule of this protein is digested. The DH value of the digested protein is 18. Calculate the number of peptide bonds that have been hydrolysed. Answer			(Total	8 marks)
(a) (i) A protein molecule contains 151 amino acids. What is the total number of peptide bonds in this molecule? (ii) A molecule of this protein is digested. The DH value of the digested protein is 18. Calculate the number of peptide bonds that have been hydrolysed. Answer	exte	nt to w	which a protein is digested is called the degree of hydrolysis (DH). The DH value may	y
bonds in this molecule? (ii) A molecule of this protein is digested. The DH value of the digested protein is 18. Calculate the number of peptide bonds that have been hydrolysed. Answer (b) What would be the DH value of a protein if it were completely hydrolysed to amino acids? Explain how you arrived at your answer. DH value			$DH = \frac{100 \times \text{Number of peptide bonds hydrolysed}}{\text{Total number of peptide bonds present}}$	
Calculate the number of peptide bonds that have been hydrolysed. Answer (b) What would be the DH value of a protein if it were completely hydrolysed to amino acids? Explain how you arrived at your answer. DH value	(a)	(i)		
Calculate the number of peptide bonds that have been hydrolysed. Answer (b) What would be the DH value of a protein if it were completely hydrolysed to amino acids? Explain how you arrived at your answer. DH value				(1)
(b) What would be the DH value of a protein if it were completely hydrolysed to amino acids? Explain how you arrived at your answer. DH value		(ii)	· · · · · · · · · · · · · · · · · · ·	
(b) What would be the DH value of a protein if it were completely hydrolysed to amino acids? Explain how you arrived at your answer. DH value				
Explain how you arrived at your answer. DH value			Answer	(1)
	(b)			?
Explanation		DΗ ν	value	
		Expl	anation	
				(2)

For more help visit our website https://www.exampaperspractice.co.uk/



Enzymes **A** and **B** digest protein. The graph shows the effect of pH on the rates of reaction of these enzymes.



(c) Pepsin is a protein-digesting enzyme found in the stomach. It has an optimum pH of 2 and is fully denatured at pH 6. Sketch a curve on the graph to show the effect of pH on the rate of reaction of pepsin.

(d) Explain why the rate of reaction of enzyme **B** is low at pH 5.

(e) Enzyme **A** is present in some washing powders used for cleaning clothes. Use the graph to suggest why enzyme **A** would be of more use in washing clothes than enzyme **B**.

(1)

(3)

(1)



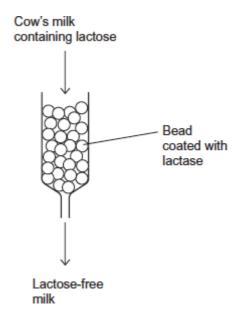
(Total 15 marks)



21

Cow's milk contains the sugar lactose. Many cats are unable to digest cow's milk because they are lactose intolerant.

Cow's milk can be made suitable for these cats by treating it with the enzyme lactase to hydrolyse lactose. This makes the cow's milk lactose-free. Beads are coated with lactase and placed in a tube, as shown in the diagram below. Cow's milk flows over the beads and the lactose is hydrolysed.



(a) Attaching lactase to the beads is a more efficient use of lactase than adding the lactase directly to cow's milk.

Suggest **three** reasons why it is more efficient to attach lactase to the beads.

1	 	 	
2			
3		 	

(3)



	The lectors from milk made after hydrolysis with lectors tested awaster than the sou	u'a mille
	The lactose-free milk made after hydrolysis with lactase tastes sweeter than the cov- containing lactose.	w S milk
	Suggest why.	
		_
		(T-1-1-F
		(Total 5 mark
the c	nciple of homeostasis is the maintenance of a constant internal environment. An inc concentration of carbon dioxide would change the internal environment and blood pH ain the importance of maintaining a constant blood pH.	
		_
		_
		_
		_
		_
		- -
		_
		- - -
[Extr	ra space]	- - -
	ra space]	- - -
[Exti	ra space]	- - -
[Extr	ra space]	- - - -
[Exti	ra space]	- - -
[Exti		- - - - (Total 3 mark
		- - - - (Total 3 marl
[Extr		- - - - (Total 3 mari
		- - - - (Total 3 marl
		- - - - (Total 3 mar
		- - - - (Total 3 mar
		- - - - (Total 3 mar
		- - - (Total 3 marl
		- - - - (Total 3 marl



(b)	Gout is a disease caused by the build-up of uric acid crystals in joints. Uric acid is produced
	from xanthine in a reaction catalysed by the enzyme xanthine oxidase.

Allopurinol is a drug used to treat gout. The diagram shows the structures of xanthine and allopurinol.

Use this information to suggest how allopurinol can be used to treat gout.						
(Extra space)						

(3)

(Total 5 marks)