

Mark Scheme

Q1.

Question number	Answer	Notes	Marks
(a) (i)	water;		1
(ii) (iii)	 prevents other larger molecules from passing; 		2
()	 to produce ammonium ions; to react with/stimulate sensor; 		3
(b)	 temperature affects enzyme; higher temperature, higher rate of reaction; more ammonium ions produced; higher reading; 	ORA for lower temperature	4
(c)	 any two from can be reused; products not contaminated; enzymes more stable; 		2
		Total 1	12 marks



Q2.

Question number	Answer	Notes	Marks
(a) (i)	 reference to platelets; fibrinogen converted to fibrin; (fibrin) mesh/crosslinks/fibres formed; 		3
(ii)	 reduced blood flow to heart/cardiac muscle / tissue/cells; less oxygen/glucose delivered; less (aerobic) respiration/energy released (by cardiac tissue); 		3
(b)(i)	size of blood clot/volume of water /aspirin solution used/temperature/shape of blood clot/width of straw;	Allow valid alternatives	1
(ii)	use water without aspirin dissolved;		1
(c)	 the more tablets dissolved the more solution collected; the more tablets dissolved the quicker the solution is collected; aspirin dissolves/acts on blood clot; size of blood clot reduced more quickly with more tablets/blood clot dissolves/breaks down faster with more tablets; Any four from: aspirin binds to enzyme/aspirin competes (with substrate) for active site; active site of enzyme changed/blocked; less/no substrate binds (to enzyme); reaction involving release of chemicals reduced/stopped; platelets do not stick together; 		Max 3
	т	otal question = 15 mar	ks

Q3.

Question number	Answer	Notes	Marks
(a) (i)	 allows protein to stay longer in stomach/gut (1) 		
	 gives time for full digestion (1) 		2
(ii)	 diet largely/entirely milk when young (1) 		
	 changes to solids when older (1) 		2
(iii)	biuret test (1)		
	add reagent to solid (1)		2
	blue to lilac/pale purple (1)		3
(b) (i)	temperature (1)		
	 volume of milk (1) volume/concentration of chymosin (1) 		3
(ii)	$\frac{238 + 232 + 241 + 229}{4}(1) = 235 (1)$		2
(iii)	not included when calculating mean (1)		1
(iv)	bubbles of different sizes/volume of CO ₂ variable (1)		1
	Total for	this question = 14 marks	

Q4.

Question number	Answer	Mark
(a)(i)	 An explanation that makes reference to the following linked points: similar/same shape to the substrate (1) therefore complementary in shape to the enzyme (1) binds to the active site (1) enzyme is lock and substrate is the key/according to lock and key hypothesis (1) 	
		3

Question number	Answer	Mark
(a)(ii)	A description that makes reference to any two of the following points: • lowers activity (of enzyme) (1) • takes longer to reach maximum rate of reaction (1) • no effect at maximum substrate concentration (1)	2



Question number	Answer	Mark
(b)	A description that makes reference to any three of the following points: this is non-competitive inhibition (1) binds at a place on the enzyme other than the active site (1) changes shape of enzyme/active site (1) substrate cannot bind to enzyme/active site (1) irreversible (1)	
		3

Question number	Answer	Additional guidance	Mark
(c)(i)	Process:	award 2 marks for correct	
	(3 900 000 000 ÷ 100) × 21 (1) = £820 000 000 (1)	final answer	
		allow £819 000 000 for	
		both marks	
			2

Question number	Answer	Mark
(c)(ii)	B A	1

Question number	Answer	Mark
(d)	A description that makes reference to the following points: immobilised enzyme bound to cardboard strip (1) urine passed over strip including the enzyme (1)	
	 glucose binds to immobilised enzyme if present converting it to hydrogen peroxide/gluconic acid (1) causes a colour change/visible colour to appear (1) 	4



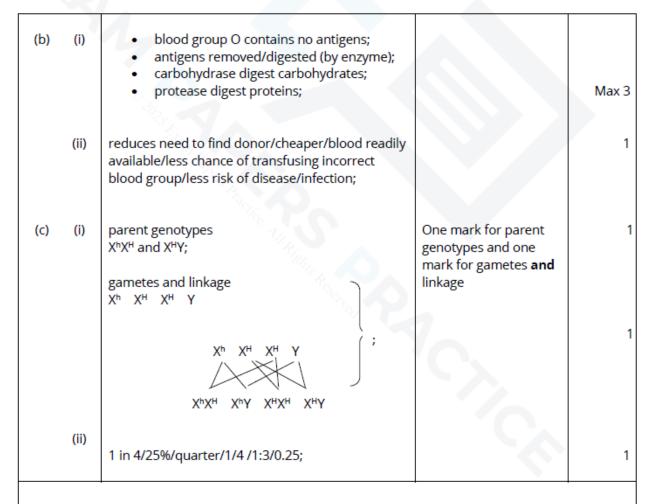
Question number	Answer PAPERS PRACTICE	Notes	Marks
(a) (i)	C; (lipase)		1
(ii)	A; (carbon, hydrogen, oxygen)		1
(iii)	C; (liver)		1
(b)	 show effect of enzyme (1) show that bile salts on their own don't break down fat (1) 	Ignore control	2
(c)	 create alkaline/optimum pH (1) for enzyme to work quickly (1) 	Allow increases/adjusts pH	2
(d)	 increase rate of digestion (1) with no bile salts change occurred after 15 minutes/digested more slowly in tube A (1) occurred after 5 minutes with bile salts present/digested more quickly in tube B (1) 		3

Question number	Answer	Notes	Marks
(a) (i)	A (amino acids); B and D are components of lipids C is a carbohydrate		1
(ii)	 add Biuret solution to test solution; mauve/lilac, protein present; remains blue, protein absent; 		3
(b) (i)	 suitable axes labels; suitable scale; correct plots; suitable curves; curves labelled; 	max 3 for bar chart accept key	5
(ii)	 denaturation; loss of shape of active site/enzyme; substrate no longer fits/decreases rate of activity; 		3
(iii)	 any three from B is much more active/works faster; over whole temperature range; higher optimum temperature for B/works fastest at 80°C; active over a greater range of temperatures; 		
	active over a greater range or temperatures,		3

Question number	Answer		Notes	Marks
(a) (i)	Conditions	Result/no/scale of bubbles		3 marks
	Room temperature	5		
	Alkali added	3		
	Acid added	2		
	3°C	1		
	70°C	0		
	no hydrogen peroxide added	0		
	two column table; headings; correct order of re	sults (high to low/ORA);		
(ii)	no repeats/only do	one once;		1 mark
(iii	results based on judgement/subjec	tive/qualitative;		1 mark
(iv	measure volume of	f oxygen; other suitable apparatus;		3 marks
(b) (i)	untreated/room temperate	ure;		1 mark
(ii)	control;	THE RIGHTS		1 mark
(iii	reduced/low enzyr not at <u>optimum</u> pH active site changed fewer enzyme subscomplexes/substra	l; d; strate	7 _C	4 marks
			Tota	al 14 marks

Q8.

Question number	Answer	Notes	Marks
(a) (i)	A (the inheritance of two different alleles, both of which are expressed)		1
	B is incorrect as one allele is not expressed in codominant inheritance C is incorrect as the ABO inheritance does not involve the inheritance of multiple alleles D is incorrect as ABO inheritances does not involve the inheritance of multiple alleles		
(ii)	I ^A I ^A and I ^A I ^O		1



Total question = 9 mark

Q9.

Questi numb		Answer	Notes	Marks
(a)	(i)	sucrase/invertase (1)		1
	(ii)	 enzyme is a protein (1) biuret test (1) purple if protein present/remains blue if absent (1) 		3
	(iii)	reusable (1)stable (1)		2
(b)	(i)	 allows glucose molecules to pass (1) prevents larger/other molecules/ blood cells from passing (1) 		2
	(ii)	 breaks down/converts glucose (1) into hydrogen peroxide (1) which activates electrode (1) more glucose results in more hydrogen peroxide (1) 		4

Q10.

Question number	Answer	Notes	Marks
(a)	 bell-shaped line (1) peak at 37°C ± 3°C (1) 		2
(b)	 increase in temperature (1) molecules have more (kinetic) energy (1) more collisions (1) more enzyme-substrate complexes formed (1) greater rate of activity (1) 	70.	5
(c)	hot (1) because optimum temperature is between 60-75°C (1)		2
	Total	for this question = 9 marks	



Q11.

Question number	Answer	Notes	Marks
(a)	DNA made from amino acids Carbohydrate made from sugar prosein made from amino acids Epid made from	Do not allow more than one line from each structure	
4	Spati made from nucleotides DNA made from nucleotides		3
(b)	D stomach;		1

(c)	(i)	3/three			1
	(ii)	not washing equipment/transfer of food containing protein into distilled water;		Allow any valid alternative	1
	(d)	Hazard	Reducing risk		
		Broken glass - cuts	Clear up breakages using a dustpan and brush/keep equipment away from edge of tables;	Ignore wear gloves	1
		Biuret reagent - irritant	Wear goggles/wash hands after use/don't use if skin is sensitive;	Ignore wear gloves	
					1
	Total question = 8 marks				

Q12.

Question number	Answer	Notes	Marks	
(a) (i)	C green vegetables and red meat		1	
(ii)	 (iron) needed for production of haemoglobin(1) (haemoglobin/red blood cell) binds to/transports oxygen(1) oxygen needed for (aerobic) respiration/energy(1) for growth of fetus(1) 		4	
(b) (i)	Any four from: take temperature of water (at the start) (1) reference to measuring mass of food(1) place food on holder(1) burn food(1) reference to time (for burning food)(1) take temperature of water after burning/once food is completely burnt/measure change in temperature(1)	Allow amount for mass Allow food is heated/set alight	4	
(ii)	Any three from: add a lid(1) reference to stirring/even heat distribution(1) insulate the copper can/move food closer (to the can/water) (1) reference to preventing heat loss(1)	Allow cover the beaker	3	
	Total for question = 12 marks			



Q13.

Question number	Answer	Notes	Marks
(a)	 4 of add powder to water (in a test tube); add Benedict's reagent/solution; heat tube in a water bath; observe any colour change/description of colour change; wear goggles; 		4 max
(b) (i)	powder Y;		1
(ii)	powder W;powder Z;		2
(iii)	powder X;		1
(iv)	protein;		1
			Total 9

Q14.

Question number	Answer	Mark
(a)	 A graph showing: vertical axis scale half grid and linear (1) lines drawn connecting points (1) horizontal axis labelled hours and vertical axis labelled grams (1) points plotted correctly (1) key for amylase Q/amylase P (1) 	
		5

Question number	Answer	Mark
(b)(i)	An explanation that makes reference to the following points: starch digested/broken down to glucose (1) therefore causes water to enter tubing (1) by osmosis (1)	3

Question number	Answer	Mark
(b)(ii)	An explanation that makes reference to the following points: • substrate/starch concentration reduced (1) • less for enzymes to digest (1)	2

Question	Answer	Mark
number	EXAM PAPERS PRACTICE	
(b)(iii)	An explanation that makes reference to the following points:	
	concentration amylase will change the rate of reaction (1)	
	pH enzyme activity changes with pH (1)	2

Question number	Answer	Mark
(c)	To ensure that the mass is not affected by water	1

Question number	Answer	Mark
(d)	Salivary glands (1) Pancreas (1)	2

Question number	Answer	Mark
(e)	A description that makes reference to three of the following points: Benedict's test (1) heat sample (1) brick red/orange colour shows presence of glucose (1)	
		3

Q15.

Question number	Answer	Notes	Marks
(a) (i)	better <u>transfer</u> of heat to water (1)		1
(ii)	 safety goggles/glasses (1) water may spit (1) use test tube holder (1) to avoid burning hands (1) 	reason linked to precaution	2
(b) (i)	• 28 x 12 x 4.2 (1) • 1411/1411.2J (1) • 1400J (1)	error carried forward full marks for correct answer	3
(ii)	 some energy lost to air/absorbed by tube/needle (1) some converted to light (1) food not fully burnt; 		3
(c)	 use same mass of food (1) same volume of water (1) use same thin tube (1) burn food same distance from tube (1) 		4



Q16.

Question number	Answer	Notes	Marks
(a) (i)	 safety glasses/goggles; care using sharp needle; care to avoid burns; 		3
(ii)	 mass of food; volume of water; distance of burning food from tube; 	reject amount	3
(iii)	any five from		
	 not all food burns to release energy; not all heat transferred to water; some lost to atmosphere; some heat lost to needle/glass tube; uneven distribution of temperature in water; because lack of stirring; 		5
(b)	any four from		
	 use of oxygen; allows complete combustion; enclosed/lid so no heat escapes; heat transfer coil so all heat is used to increase temp of water/transferred to water; stirrer to distribute heat; 		4
	Why.	Total	15 marks