

12.2 Nuclear Physics

Question Paper



Exam Papers Practice

To be used by all students preparing for DP IB Physics HL Students of other boards may also find this useful



Question 1

Which of the following statements about the nuclear radius is incorrect?

- A. As more nucleons are added to a nucleus the radius increases
- B. The radius of nuclei depends on the atomic number, Z of the nucleus
- C. The exact relationship between the radius and nucleon number can be determined from experimental data
- D. The radius of the nucleus can be calculated using an equation which involves the constant of proportionality, R₀

[1 mark]

Question 2

The graph shows the scattering of electrons due to diffraction by oxygen-16 nuclei.



Which point on the graph represents the first minimum?

[1mark]

Question 3

What do the deviations from the Rutherford scattering give evidence for?

- A. The nuclear radius
- B. The charge on the nucleus
- C. The electrostatic force
- D. The strong nuclear force

[1mark]



Question 4

What is the radius of cobalt-27?

- A. 3.6 fm
- B.10.8 fm
- C. 27 fm
- D. 32.4 fm

[1 mark]

Question 5

Which graph shows how intensity l varies with angle θ when electrons are diffracted by a nucleus?



Question 7

Which of the following graphs of the number of undecayed nuclei N over time has the smallest decay constant?





When alpha particles of speed v are directed at nuclei with atomic number A, scattering occurs. Small deviations from the predicted intensity of the scattered particles are observed.

Which change in nucleon number and energy will give the greatest deviation from the Rutherford scattering predictions?

	Atomic number	Speed
Α.	decrease	increase
В.	decrease	decrease
C.	increase	increase
D.	increase	decrease
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Question 10

A radioactive element has a decay constant λ . At time, t = 0 the number of nuclei of this element is N₀. These are linked by the equation $N_t = N_0 e^{-\lambda t}$ where N_t is the number of nuclei expected at time t.

What number of nuclei would you expect to remain after 1 second?

A.0 B. $N_0(1 - e^{-\lambda})$ C. $N_0e^{-\lambda}$ D.1

[1 mark]